

## • Artificial Satellites

in the book (their respective fields of science are given in parentheses): Mikhnevich, V.V., (temperature measurement at the altitude of 80-85 km); Astanovich, I.S. and Kaplan, S.A., (satellite's speed of re-entry); Okhotsimskiy, D.Ye., Eneyev, T.I. and Taratynova, G.P., (deceleration due to the resistance of air); Shternfel'd, V. and Vorontsov, V., (space ships); Kondratyuk, Yu.V., (Interplanetary flight); Barbarov, N.A., (rocket launching from aircraft); Tsander, F.A. (finned rockets); Yegorov, V.A. (motion of satellites); Pokrovskiy, A.V., (experimental flights to the altitude of 110-200 km); Orlov, V.S. and Levin, B.Yu., (study of meteorites); Karpenko, A.G. and Lidov, M.L., (conditions of temperature in the Earth satellites); Vavilov, V.S., Malovetskaya, V.M., Galkin, G.N., Landsman, A.P., (improvement of silicon photo elements used as a source of power in artificial satellites); Molchanov, P.A., (automatic radio communication); Gringauz, K.I. and Zelikman, M.Kh., (study of the concentration of positive ions along the satellite's orbit); Istomin, V.G., (study of ionized layers of the atmosphere); Dubrovin, (determination by radio of the culmination point of a satellite); Yatsunskiy, I.M., (geodesic research); Danilin, B.S., Repnev, A.I., Shvidkovskiy, Ye.G., (precise methods of density and pressure measurement); Fesenkov, V.G., (study of meteorites);

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Poloskov, S.M., Nazarova, T.N., (study of the concentration of the interplanetary dust on the surface of the Earth); Ginsburg, V.L., (study of the luminosity of the meta-galactic); Mandel'shtam, S.L., Yefremov, A.I., (study of the ultra-violet rays); Vernov, S.N., Ginsburg, V.L., Kurnosova, L.V., Razorenov, L.A., Fradkin, M.I., (research on primary cosmic radiation); Vavilov, Cherenkov, (study of charged particles moving at a speed surpassing the speed of light in the medium in which it is traveling); Logachev, Yu.I., Chudakov, A.Ye., Shafer, Yu.G., (study of cosmic radiation); Gaysler and Pan'kov, (Poland, changing the astroid Hermes to a satellite of the Earth). There are no references.

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AVAILABLE: Library of Congress

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SHTERNFEL'D, Ario Abramovich

"Artificial Satellites of the Earth,"

Published by the State Publishing House of Technical and Theoretical Literature,  
Moscow, V-71, B. Kaluzhakaya, 15. Pages 1-180, 1958

Trans . JPRS Rpt 191

SHTERNFELD, A.

Artificial satellites. Tekhn.mol. 26 no.2:5-7 '58. (MIRA 11:2)  
(Artificial satellites)

29(0)

PHASE I BOOK EXPLOITATION

SOV/3287

- Shternfel'd, Ario Abramovich, Winner of the International Prize for the Promotion of Astronautics

Ot iskusstvennykh sputnikov k mezhplanetnym poletam (From Artificial Satellites to Interplanetary Flights) 2d ed., rev. and enl. Moscow, Fizmatgiz, 1959. 202 p. 75,000 copies printed.

Ed.: I. Ye. Rakhlin; Tech. Ed.: N. Ya. Murashova.

PURPOSE: This booklet is intended for the general reader interested in space exploration and travel.

COVERAGE: This booklet provides a generalized account of extra-terrestrial exploration as of October 1959. The author discusses both Soviet and non-Soviet space vehicles, satellites, and experiments. He treats lunar probes and the prospects of a man-in-space project. The possibilities of interplanetary travel are also discussed. TASS news bulletins on the launching of the second and third Soviet space rockets are included. The Appendix contains technical specifications of artificial earth satellites.

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No personalities are mentioned. No references are given.

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3/16/60

Shternfel'd, Ario Abramovich

Soviet Space Science, by Ari Shternfeld. 2nd. Rev. ED. New York, Basic Books, 1959.

XXII, 361 P. Illus., Charts, Diagr., Tables.

Translated by the Technical Documents Liaison Office, Wright Patterson Air Force Base, Ohio, from the original Russian title: Iskusstvennyye Sputniki. Moskva, 1958.

Also published under title: Artificial Satellites. Wright-Patterson Air Force Base, Ohio, 1958.

29(0)

AUTHOR:

Shternfel'd, A., Holder of the  
International Prize for the  
Promotion of Astronautics

SOV/29-59-4-5/26

TITLE:

The Moon of the Big Moon (Luna bol'shoy Luny)

PERIODICAL:

Tekhnika molodezhi, 1959, Nr 4, p 4 (USSR)

ABSTRACT:

The author reports here on the possibility of creating an artificial moon satellite. This would make it possible to observe the moon surface in smallest details on the earth by means of a television camera. It does not appear so very difficult to send a rocket to the other side of the moon. A far bigger problem is that of controlling the propulsion in a way as to apply to the rocket the required speed and direction. The smallest deviation from the calculated speed would cause the rocket either to crash on the moon or to pass it at too great a distance. To avoid any damage, the entire trajectory should be under control. At any rate, it will hardly be possible for a rocket to approach safely the moon by less than 200 km. According to calculations this is the distance at which the rocket could be transformed into a moon satellite. Furthermore, its horizontal speed should be such as to prevent

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The Moon of the Big Moon

SOV/29-59-4-5/26

the satellite by the mechanical onward movement from crashing on the moon. This speed is 1590 m/sec, i.e. about five times less than the speed of the satellite revolving around the earth. By the said speed the satellite would revolve around the moon in 2 hours 7 minutes on an orbit of 12,177 km. It would not fly for more than 45 minutes 15 seconds in the shadow of the moon. The mountain peaks on the moon are about as high as Mount Everest, i.e. about 9 km. and that means that the satellite should not fly lower than 10 km above the moon surface. To lower the satellite from a distance of 200 km to 10 km, its flying speed should be slowed down by 41.5 m/sec, in which case it would deviate from its orbit and approach the moon surface. The "drop" in an elliptical curve would take 1 hour 2 minutes. The speed would rise to 1719 m/sec and the satellite would soar up again to 200 km altitude. The time it would be in the proximity of the moon surface would be too short as to permit accurate observations. For this reason it would be necessary to put it into an orbit. This would be made possible by decreasing its speed of 1719 m/sec by 45 m/sec at 10 km altitude. Satellites flying at an altitude of 200 - 10 km could revolve around the moon for about 308-363 times

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SHTERNFEL'D, A.A., laureat Mezhdunarodnoy premi pooshchreniya  
astronavitiki (Moskva)

Flights into space. Nauka i zhyttia 10 no.2:14-17  
F '60. (MIRA 13:6)  
(Astronautics)

SHTERNFEL'D, A.A., laureat Mezhdunarodnoy pooshchritel'noy premii po  
astronavtike (Moskva)

Space flight to Venus. Nauka i zhyttia 10 no.7:18-21  
J1 '60. (MIRA 13:7)  
(Space flight to Venus)

SHTERNFEL'D, ARIO ABRAMOVICH

FROM MAN-MADE SATELLITES TO INTERPLANETARY FLIGHTS.  
[TRANSLATION] PREPARED BY: TECHNICAL DOCUMENTS  
LIAISON OFFICE, MCLTD, WRIGHT-PATTERSON AFB, OHIO,  
1961

210L. ILLUS., DIAGRS., TABLES (ITS: MCL-1303/1 #2)  
TRANSLATED FROM THE ORIGINAL RUSSIAN: OT ISKUSST-  
VENNYKH SPUTNIKOV K MEZHPLANETNYM POLETAM. 2D. REV.  
ED. MOSCOW, 1959

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S/029/61/000/011/003/004  
D036/D113

3,2300

AUTHOR: Shternfel'd, A., Winner of the International Incentive Prize  
For Astronautics

TITLE: Twenty-four hours in space opens the way to the planets

PERIODICAL: Tekhnika molodezhi, no. 11, 1961, 14-15

TEXT: The author discusses space flights which could be completed in 25 hours 18 minutes, the duration of German Titov's space flight. During this time, a rocket fired vertically from the Earth's surface at a launching speed of 10.8 km/sec, its engines subsequently being switched off, would reach a height of 90,000 km. To carry out such a flight and then return to Earth, the rocket would ascend for 12 hours 39 minutes, developing a speed of 10.7 km/sec; reaching a maximum altitude of 81,000 km, it would stop for an instant and then descend, reaching the Earth 25 hours and 18 minutes after launching time. In 25.3 hours a spaceship could fly around the Earth along a huge closed ellipse with a major axis of 87,400 km and a minor axis of 23,100 km. The engines would be switched off along the perigee at an altitude of 200 km, after which the ship would coast, gradually gaining

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Twenty-four hours in space ...

altitude, until after 12 hours 39 minutes it would reach its apogee 74,400 km above the Earth's surface. The spaceship would then begin to descend, and after the same period of time would return to the Earth's atmosphere at an altitude of 200 km. The launching speed of the spaceship for such a flight would be 10.2 km/sec. Cosmonauts aboard a stationary satellite would be able, under certain conditions, to "jump aboard" another satellite flying along a circular orbit with a radius of 43,700 km and a circling time (in respect to the stars) of 25 hours 18 minutes - a period during which Titov suffered no ill effects due to weightlessness. During their stay on the stationary satellite, the cosmonauts would be able to create artificial gravity aboard by rotating the craft about its center of gravity. A huge figure eight would be produced by joining all the points on the Earth's surface from which such a satellite (i.e. one having an orbital radius of 43,700 km and an orbital plane passing through the poles of the Earth, and a circling time of 25 hours 18 minutes) would be successively visible at its zenith. In the case of an equatorial satellite, this projection would have the form of a straight line. The speed and direction of an equatorial satellite with a circling time of 25.3 hours, as seen by an observer at the equator, would vary in accordance with whether the direction of the satellite's motion

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Twenty-four hours in space ...

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coincided with that of the Earth's axial rotation or not. In the former case it would appear to be travelling very slowly, and to get into orbit from a stationary satellite would require an initial speed of only 27 m/sec. In 25 hours it would be possible to fly to the Moon. This would require a launching speed of about 12 km/sec, only 8% more than the minimum launching speed, when the flight would last 5 days and nights. In the case of a 25-hour-flight, the space ship would fly along a hyperbolic trajectory and reach the Moon's orbit at a speed of 4.4 km/sec; for the 5-day-flight the trajectory would be semi-elliptical and the apogean speed only 0.2 km/sec. As the ship would move towards the moon at 2.4 km/sec due to lunar attraction, the speeds which would have to be counteracted for landing on the moon, would be 2.6 km/sec for the 5-day-flight and 6.8 km/sec for the 25-hour-flight. Calculations show that it would be necessary to brake at a speed of 2.41 km/sec in the former case and 5.01 km/sec in the latter. If the rocket's exhaust velocity is 3 km/sec, then the weight of fuel required for braking would be 1.23 times greater than the weight of the rocket in the first case and 4.31 times greater in the second case. In 25 hours 18 minutes, an artificial satellite could fly once around the Sun in an orbit

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Twenty-four hours in space ...

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with a radius of 3,040,000 km. It would then be 2,340,000 km from the Sun's surface, at which distance the Sun's rays would be 19 times less intense than near its surface. If the satellite were shaped like a greatly extended cylinder with a hemisphere at each end, if its axis were oriented parallel to the Sun's rays, if the hemisphere facing the Sun were made to reflect back almost all the rays and the rest of the skin were made of heat-reflecting material, then the internal temperature of the cabin would not be too high for the cosmonaut. Most of the satellite's surface would be in the shade. In 25 hours 18 minutes, an artificial satellite could fly once around the largest comet known to man - that observed in 1818, with a mass of  $2.10^{13}$  tons and a nucleus 20 km across - in a circular orbit with a radius of 59 km. The speed of such a satellite would be 15.5 km/hr. In conclusion, the author remarks that the longer man will be able to stay in the cosmos, the greater the number of possible cosmic flights. There are figures.

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SHTERNFEL'D, A.A., laureat Mizhnarodnoi zaokhochuval'noi premii z astronavtiki

Conquering space. Nauka i zhyttia 11 no.5:29-32 My '61.  
(MIRA 14:7)

1. Pochesniy chlen Lotarings'kogo vchenogo Tovaristva (Moskva).  
(Astronautics)

SHTERNGAS, Ya.; AFANAS'YEVA, K.

Polyvinyl chloride linoleum. Stroi. mat., izdel. i konstr.  
2 no.7:13-14 J1 '56. (MLRA 9:10)

1. Glavnyy inzhener Mytishchinskogo silikatnogo zavoda  
(for Shterngas).  
(Linoleum) (Ethylene)

SHTERNGAS, Ya., inzh.

Possibilities of using higher pressure in autoclaves. Stroi. mat.

4 no.9:14-17 S '58.

(MIRA 11:10)

(Autoclaves)

ARISTOV, F.M.; SHTERN GAS, Ya.S.

Output of linoleum has increased 50%. Stroil. mat. 10 no.3:  
6-7 Mr '64. (MIRA 17:6)

1. Direktor Khlyupinskogo zavoda linoleuma (for Aristov).
2. Glavnyy inzh. Khlyupinskogo zavoda linoleuma (for Shtern gas).

SHTERNGERTS, A.N.

Ophthalmologists in Sweden; remarks of a physician. Vest. oft. 71  
no.2:50-52 Mr-Ap '58. (MIRA 11:4)

1. Glaznoye otdeleniye bol'nitsy No.3 g. Gorlovki.  
(OPHTHALMOLOGY  
in Sweden)



SHTERNGERTS, I.S.

Staining of trichomonas. Akush. gin. no.3:67 May-June 1953. (GIML 25:1)

1. Of Sverdlovsk Scientific-Research Institute for the Care of Mother and Child (Director -- R. A. Malysheva).

Central Inst. of Microbiology and Microbiology of NKZMAYA (People's Commissariat  
Public Health), (1955)

The corn and wheat grain excretion.".

Ann. Microbiol., Bacteriol., & Immunol., No. 1-2, 1955.

GIL'DIN, S. R., SHETERNGOL'D, YE. YA., ASIMARIN, I. I., ZHDANOVA, L. D.,  
ZVAGEL'SKAYA, V. N., KALININA, YE. F., LOSKUTOVA, N. N., PYZHOVA, M. E., AND  
SLAVINA, A. M.

Further Observations on the Effectiveness of Subcutaneous Vaccination Against  
Dysentery

Shows that the epidemiologic effectiveness of subcutaneous vaccination  
against dysentery is very low and has no advantages over the enteral method.  
(RZhBiol, No. 7, 1955) Vopr. Kravevoy Patologii AN UzSSR, 3, 1953, 51-52

SO: Sum. No. 744, 8 Dec 55 - Supplementary Survey of Soviet Scientific  
Abstracts (17)

ZVAGELSSKAYA, V.N.; SHTERNAL'D, Ye.Ya.

Finds of *Sp. latyshevi* in rodents in Uzbekistan. Med. paras.  
i paraz. bol. no.2:180 Ap-Je '54. (MLRA 7:8)

1. Iz Tashkentskogo instituta epidemiologii i mikrobiologii.  
(SPIROCHAETA,  
\*latyshevi, in rodents)  
(RODENTS,  
\*Spirochaeta latyshevi in)

SHTERNGOL'D, Ye.Ya.

Role of the reticulo-endothelial system in the mechanism of immunity  
to typhus fever. Vop.kraev.pat. no.4:88-98 '54. (MLRA 9:12)

(IMMUNITY) (TYPHUS FEVER)

(RETICULO-ENDOTHELIAL SYSTEM)

SHTERNGOL'D, Ye.Ya.

Immunological reactivity of animals in experimental typhus fever in connection with the different functional states of their nervous system. Vop.kraev.pat. no.4:99-107 '54. (MLRA 9:12)

(TYPHUS FEVER--PREVENTIVE INOCULATION)

(NERVOUS SYSTEM)



KHODUKIN, N.I.; SHTERNGOL'D, Ye.Ya.; SHLEYKHEN, E.I.; ZVAGEL'SKAYA, V.N.

Experience in the preparation of vaccine against Q fever. Zhur.  
mikrobiol.epid. i immun. 27 no.7:22-23 Jy '56. (MLRA 9:9)

1. Iz Tashkentskogo instituta vaktsin i syvorotok.  
(Q FEVER, prev. and control  
vaccine prep. from spleens of white mice)  
(SPLEEN  
prep. of Q fever vaccine from spleen of white mice)



SHTERNGOL'D, Ye.Ya.

Producing a non-reactogenic vaccine against Q fever. Zhur. mikrobiol.,  
epid. i immun. 32 no.9:12-18 S '61. (MIRA 15:2)

1. Iz Tashkentskogo instituta vaktsin i syvorotok.  
(Q FEVER) (VACCINES)

SHTERNGOL'D, Ye. Ya.

Use of microagglutination with R. burneti in the diagnosis of Q fever. Zhur. mikrobiol., epid. i immun. 32 no. 10: 136 0'61.  
(MIRA 14:10)

1. Iz Tashkentskogo instituta vaktsii i syvorotok.  
(Q FEVER) (SERUM DIAGNOSIS)

SHTEPNOL'D, Ye.Ya.

Reactivity of the RES elements of white mice during intraperitoneal  
infection with exanthematous typhus. Trudy TashNIIVS 6:175-181 '61.  
(MIRA 15:11)

(TYPHUS FEVER)

(RETICULO-ENDOTHELIAL SYSTEM)

SHTERN GOL'D, Ye.Ya.; SHLEYKHER, E.I.; UMIDOVA, L.SH.

Immunological effectiveness of Q vaccine. Trudy Tash. NIIVS  
5:47-52'62. (MIRA 16:10)  
(Q FEVER — PREVENTIVE INOCULATION) (IMMUNITY)

AUTHORS: Faygenbaum, D.S., and Shternin, L.A., Engineers 135-9-19/24

TITLE: Condenser Contact Welding Machine "MTK-0,1" (Kondensatornaya machina dlya kontaktnoy svarki "MTK-0,1")

PERIODICAL: "Svarochnoye Proizvodstvo", 1957, # 9, p 36-37 (USSR)

ABSTRACT: The condenser contact welding machine "MTK-0,1" was designed for welding ferrous and non-ferrous metals of 0.02 to 0.15mm thickness. It is a table model and can work with a stationary welding head as well as with changeable welding tools and is suited for welding in assembling instruments and radio equipment. The design of the machine is described in detail and illustrated by a photograph and an electric diagram. The advantages of electrolytic condensers are discussed. Condensers "KЭ -2H" were found to be the most advantageous. These latter are extensively employed in mass-produced radio sets (in feeding filters of broadcasting receivers and television sets).

The article contains 1 photograph and 2 diagrams.

ASSOCIATION: VNIIESO

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A006/A002

1.2300 also 2508

AUTHORS: Shternin, L. A., and Bol'shev, S. A., Engineers

TITLE: Friction Welding of Parts at the "Elektroapparat" Plant

PERIODICAL: Svarochnoye proizvodstvo, 1960, No. 9, pp. 37-38

TEXT: 1 The MCT-2 and the MCT-31 (MST-2 and MST-31) friction welding machines were brought into use at the Leningrad "Elektroapparat" Plant. Their technical data are given in Table 1. The introduction of the machines was performed with the assistance of VNIIESO where they had been designed and manufactured. The machines are intended for small-scale production. Ordinary three-jawed holding devices of 165 and 250 mm respectively are used. The reduction forces in welding are transmitted to supports limiting the length of the parts to be welded to 380 mm on the MST-2 and to 1,000 mm on the MST-31 machine. The welding cycle is automated. The parts to be welded are, as a rule, connections of a rod with a plate. Flat parts are mounted on a device which is fixed to one of the holders, thus reducing the operational time. Welding of levers, rollers, shafts, pins, insertion pieces and strips can now be performed by friction. Raised labor efficiency, reduced consumption of metal, electrodes

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S/135/60/000/009/012/015

A006/A002

Friction Welding of Parts at the "Elektroapparat" Plant

and electric power, and improved work conditions prove the efficiency of the new method. Welding conditions are given in Table 2. There are 3 figures and 2 tables.

ASSOCIATION: VNIIESO-Shtermin, L. A., "Elektroapparat" Plant-Bol'shev, S. A.

Card 2/2

S/110/61/000/001/014/023  
E194/E455

AUTHORS: Shternin, L.A., Engineer, Prokof'yev, S.N., Engineer,  
Orlov, Ya.M., Engineer and Kobyl'nitskaya, M.I., Engineer

TITLE: The Introduction of Friction Welding of Copper Current-  
Conducting Parts

PERIODICAL: Vestnik elektromyshlennosti, 1961, No.1, pp.44-45

TEXT: This article describes experience of using a friction welding machine type MCT-6 (MST-6) for friction welding of a small copper assembly. In the old method of construction, a copper pin 12 mm diameter was turned down at one end to fit a brass washer and was soldered to a strip of copper 2 mm thick. Friction welding was the most suitable for such parts, as arc welding could not be used. The machine type MST-6 has a motor of 2.8 kW, the spindle is driven at 4000 rpm and an axial force of 50 to 1000 kg can be applied pneumatically. The welding time can be controlled within the range 0.5 to 2.5 sec, and the complete cycle has a duration variable between 5 and 15 sec. The machine automatically loads the pins into the pressure device of the spindle, brings the strip up to the spindle, makes the weld and discharges the welded products.

Card 1/2



S/110/61/000/001/014/023  
E194/E455

The Introduction of Friction Welding of Copper Current-Conducting Parts

The parts are carried on a rotating table with eight positions. Pneumatic drive is used to turn the table. Welding can be effected with very little distortion of the parts. The use of the machine has simplified production of the parts; there is no need to make the brass washers, to roll the parts together or to clean them after soldering. By use of the machine, the standard time for making the parts was reduced from 1.6 to 0.25 hours per hundred. The economy of wages was 6.95 roubles per 100 parts. The properties of the finished parts are improved. It is necessary that the surfaces of all the parts should be equally clean. This is achieved by etching in a mixture of sulphuric and nitric acids, followed by water washing and compressed-air drying. There are 2 figures.

SUBMITTED: June 14, 1960

Card 2/2

1.2310

2408 1573

28984 S/135/61/000/011/006/007  
A006/A101

AUTHORS: Shternin, L. A., Prokof'yev, S. N., Engineers

TITLE: Friction welding of aluminum with steel and copper

PERIODICAL: Svarochnoye proizvodstvo, no. 11, 1961, 30-32

TEXT: Information is given on results of experiments made for the purpose of determining the basic parameters of conditions for friction welding AD-1 (AD-1) aluminum rods, 30, 40, and 50 mm in diameter with grade CT.3 (St.3) steel and M1 copper. Aluminum was friction welded with steel on a MCT-31 (MST-31) machine; the rotation of the welded blanks varied from 230 to 1,000rpm; axial forces of up to 20,000 kg were developed. The aluminum blank was clamped in a steel mandrel eliminating the heat and preventing its free deformation during welding. The blank protruded from the mandrel to a given length depending on the diameter of the specimen. The quality of weld joints was determined from the bending angle, when the welded specimens were subjected to tensile and shearing tests. It was found that the speed of relative rotation affected considerably the quality of welds; it should not be less than 760 rpm when welding 30 mm diameter blanks. Specimens of 40 mm diameter were tested, to

Card 1/3

28984 S/135/61/000/011/006/007  
A006/A101

Friction welding of aluminum with steel ...

determine the dependence of the bending angle of the welded joint on the specific pressure at heating and peening and rotation speed of 760 rpm. Specific heating pressure should not be below  $5 \text{ kg/mm}^2$ . Tests with 30 mm diameter specimens show that at this pressure the specific peening pressure does not affect the weld quality, which remain satisfactory at both constant or increased pressure. Tests with 50 mm diameter steel specimens welded with Al did not show fracture resistance of all the specimens at  $180^\circ$  bending; however, in a number of cases the results were satisfactory. Ultimate strength of the butt metal was  $10 \text{ kg/mm}^2$  against  $8.5 \text{ kg/mm}^2$  of the base metal; it was  $7.5 - 8.2 \text{ kg/mm}^2$  in the shearing tests. Microhardness corresponded to that of the base metal. Friction welding of 20 mm diameter aluminum with copper was also performed on a MST-31 machine. To remove case hardness the copper surface was machined and annealed at  $600 - 700^\circ\text{C}$  for 30 minutes. Tests showed that in all cases, excepted when the specimens were welded at  $2 \text{ kg/mm}^2$  specific heating pressure, the failure occurred in the aluminum remote from the butt. It was found that welds produced by the described method show satisfactory qualities. The main features distinguishing friction welding of aluminum with copper or steel from other metal combinations are: 1) the aluminum butts must be carefully cleaned; 2) the blanks should be fastened with steel mandrels; 3) the gauged length of the aluminum blank

Card 2/3

28984 S/135/61/000/011/006/007  
A006/A101

Friction welding of aluminum with steel ...

must be carefully observed; 4) copper blanks must be machined, annealed and cleaned; 5) high peening pressure during the welding of copper with aluminum promotes apparently the destruction and removal of brittle components, thus raising the quality of welds. There are 4 tables and 3 figures.

ASSOCIATION: VNIIESO

X

Card 3/3

32774

S/135/62/000/001/004/007  
A004/A101

1 2310 1573

AUTHORS: Shternin, L.A., Komarcheva, E.S., Val'ter, I.G., Engineers

TITLE: Friction welding in the manufacture of turbo-compressors

PERIODICAL: Svarochnoye proizvodstvo, no. 1, 1962, 14 - 16

TEXT: The authors analyze the results of technological investigations of the friction welding of austenitic 3X 572 (EI572) steel to pearlitic steels of the OXM (OKhM) and 40 X (40Kh) grades. They describe the welding conditions and heat treatment and present the results of mechanical testing of the welding joints. The investigations to study the possibility of using friction welding in the manufacture of turbo-compressor runners were carried out by VNIIESO and TsNIDI. The 40Kh grade steel was subjected to preliminary heat treatment: oil-hardening at 840°C, tempering at 550°C (with subsequent water cooling) - while the OKhM steel was welded as delivered. The austenitic steel blanks were produced by investment pattern casting and subjected to the following heat treatment: austenizing 1,160 - 1,180°C with water quenching, dispersion hardening at 750°C with 15 hours holding and air-cooling. Specimens 16, 20 and 28 mm in diameter were welded. The chemical and mechanical properties of the steels are shown in Card 1/3

32774

S/135/62/000/001/004/007

A004/A101

Friction welding ...

a number of tables. The major part of the welding operation was carried out on the MCT -31 (MST-31) machine, developing an axial stress of up to 14,000 kg, while the relative rotation speed of the blanks being welded amounted to 1,000 rpm. The authors give a description of the welding conditions and point out that an analysis of the results obtained at different welding conditions showed that the notch toughness of the welding joint was stable only under the condition of using a specific peening force equal to 21 kg/mm<sup>2</sup>. In this case, the specific heating stress amounted to 6 kg/mm<sup>2</sup>. A batch of specimens friction-welded under these conditions, was subjected to mechanical tests the results of which are shown in a table. Moreover, fatigue strength tests of the welded specimens with alternating loads were carried out on the BY -8 (VU-8) machine at the Leningradskiy politekhnicheskii Institut im. Kalinina (Leningrad Polytechnic Institute im. Kalinin). In the tensile and endurance tests all welding joints were destroyed along the EI572 steel base metal. The authors present a number of microsections, showing the microstructure of the welded specimens after etching. The analysis of the mechanical and metallographic investigations reveals that the friction welding of austenitic steel to pearlitic steel yields a dependable joint with a sufficiently high strength, exceeding in some cases the indices of welded austeni-

Card 2/3

32774

S/135/62/000/001/004/007  
A004/A101

Friction welding ...

tic steels. The equipment existing and being under construction at present for friction welding makes it possible to weld turbo-compressor runners with shafts up to 50 mm in diameter. There are 6 figures and 5 tables.

ASSOCIATIONS: VNIIESO (Shternin, L.S., Komarcheva, E.S.); TsNIDI (Val'ter, I.G.)

X

Card 3/3

SHTERNIN, L.A., inzh.; KOMARCHEVA, E.S., inzh.; VAL'TER, I.G., inzh.

Friction welding in the manufacture of turbocompressors. Svar.proizv.  
no.1:14-16 Ja '62. (MIRA 15:3)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut elektrosvarochnogo  
oborudovaniya (for Shternin, Komarcheva). 2. TSentral'nyy nauchno-  
issledovatel'skiy dizel'nyy institut (for Val'ter).  
(Compressors--Welding)(Steel--Welding)



GINZBURG, S.K., inzh.; PROKOF'YEV, S.N., inzh.; SHTERNIN, L.A., inzh.

Conditions for the formation of a resistant joint in the  
friction welding of aluminum with steel. Svar. proizv.  
no.12:12-14 D '62. (MIRA 15:12)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut  
elektrosvarochnogo oborudovaniya.  
(Aluminum--Welding)  
(Steel--Welding)

ACCESSION NR: AP4040701

S/0135/64/000/006/0023/0024

AUTHORS: Vill', V. I. (Candidate of technical sciences); Komarcheva, E. S. (Engineer); Shternin, L. A. (Engineer)

TITLE: Friction welding of thin-wall pipes made of aluminum alloys

SOURCE: Svarochnoye proizvodstvo, no. 6 (630), 1964, 23-24

TOPIC TAGS: welding, pipe, thin-walled pipe, aluminum alloy, steel 1Kh18N9T, aluminum AD1, aluminum AMts, welder MST31

ABSTRACT: Butt-welding of pipes with the ratio  $D/\delta = 25-30$  often produces deformation and lowers thermal properties. To avoid this, a new method was developed for welding thin-wall pipes different metals with different thermal properties (such as steel and aluminum). This improved friction-welding technique resulted in higher quality of welds, localized heating, small power consumption, and the even distribution of temperature along the welding surface. A serious obstacle in the practical application was the initial ellipticity of pipes and their off-axial alignment in the welder. These shortcomings were eliminated by the design of a special device shown in Fig. 1 of the Enclosure. Here two cylindrical plugs (1 and 2) were fitted into the pipes; a cylindrical rod (3) freely entered

Card 1/3

ACCESSION NR: AP4040701

the bearing (4) which was fixed in the plug (1). The guide placing the rod in the bearing secured an accurate axial alignment of the details; it did not prevent their free rotation before and during welding. Plugs fitting tightly into the pipes eliminated their ellipticity. Experiments were performed with steel 1Kh18N9T and aluminum alloys AD-1, AMts in a MST-31 welder. Brittle interlayers were eliminated, destroyed, or removed in the course of friction welding by the low rate of heating which slowed down the diffusive processes, and by forging-pressures. Orig. art. has: 1 table and 4 figures.

ASSOCIATION: VNIIESO

SUBMITTED: 00

ENCL: 01

SUB CODE: MM

NO REF SOV: 000

OTHER: 000

Card 2/3

ACCESSION NR: AP4040701

ENCLOSURE: 01

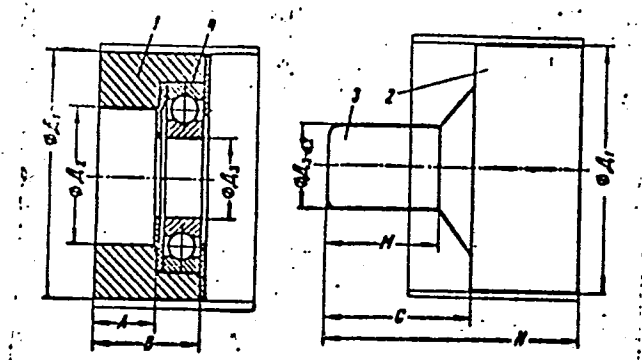


Fig. 1. Device for friction welding of thin-wall pipes.

Card 3/3

SNTLADIN, L.A., Inc.

Friction welding of thin-walled aluminum alloy pipe. Svar.  
proizv. no.6:23-24 Je '64 (MIRA 18:2)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut elektre-  
svarenchnogo oborudovaniya.

L 49459-65 EPA(s)-2/EPR/EWP(k)/EWP(z)/EWA(c)/EWT(m)/EWP(b)/T/EWA(d)/EWP(v)/EWP(t)  
Pf-4/PS-4 IJP(c) MJW/JD/HM  
ACCESSION NR: AP5007347 S/0125/65/000/003/0011/0013 60  
34  
B

AUTHOR: Shternin, L. A.

TITLE: Friction welding of unlike metals

SOURCE: Avtomaticheskaya svarka, no. 3, 1965, 11-13

TOPIC TAGS: friction welding, aluminum welding

ABSTRACT: The results are reported of an experimental study of the phenomena transpiring in the butt zone of two pieces of dissimilar metals being friction-welded. AD1, 20-mm diameter aluminum and St3 low-carbon steel specimens were studied at relative rotation speeds of 750, 1000, 1500, 3000, 6000 rpm. Specific pressure was 1-5 kg/mm<sup>2</sup>. It was found that: (1) The sustained temperature of the contact, during the welding process, is independent of the relative-rotation speed and is close to the melt point of the like metals or the lower-melt point in the case of unlike metals (low-carbon steel, 1200C; steel-

Card 1/2

L 49459-65

ACCESSION NR: AP5007347

aluminum, 550C; steel-copper<sup>21</sup>, 950C; steel-titanium<sup>21</sup>, 1200C); (2) The sustained contact temperature decreases with increasing specific pressure during the heating period; (3) The friction welding of AD1 aluminum with various steels offers no difficulty; however, Mg<sup>21</sup>, Cu-, Zn<sup>21</sup>, Si<sup>21</sup>-bearing aluminum alloys do not produce satisfactory welded joints with steel. Orig. art. has: 4 figures, 1 formula, and 1 table.

ASSOCIATION: VNIESO

SUBMITTED: 27Jun64

ENCL: 00

SUB CODE: MM

NO REF SOV: 006

OTHER: 000

Card 2/2 CC

SHTERNIN, L.A.

Friction welding of dissimilar metals. Avtom. svar. 18 no.3:11-13  
Mr '65. (MIRA 18:6)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut elektrosvarochnogo oborudovaniya.



1ST AND 2ND GROUPS																										3RD AND 4TH GROUPS																									
PROCESSING AND PROPERTY INDEX																																																			
<p>Concentrated solutions. I. Electric conductivity, viscosity and density of molten <math>\text{NH}_4\text{Ag}(\text{NO}_3)_2</math> and its concentrated solutions. M. S. Skanavi-Grigor'eva and E. B. Shternin. <i>J. Gen. Chem.</i> (U.S.S.R.) 5, 799-800 (1935).—The equiv. cond. of molten <math>\text{NH}_4\text{Ag}(\text{NO}_3)_2</math> at 110°, 112°, 113°, 115°, 118°, 120° and 123° was found to be 0.134, 0.142, 0.150, 0.158, 0.168, 0.174 and 0.187 reciprocal ohms, resp. Its viscosity, <math>1/t^*</math> mm., at 110°, 120° and 124°, was 12.05, 10.62 and 9.95, resp., where <math>t</math> is time of flow of fused salt and <math>t^*</math> mm. is time of flow of <math>\text{H}_2\text{O}</math> at 100°, through the</p> <p>Outward viscometer. Elec. cond., viscosity and d. of solns. of this salt were also studied in concns. from 41.86 to 100% at temps. of 25–112°. Conc. solns. of <math>\text{NH}_4\text{Ag}(\text{NO}_3)_2</math> represent very complicated systems. No explanation is given of the nature of curves representing variations of elec. cond., viscosity and d. with concn. or temp.</p> <p>S. L. Madorsky</p>																																																			
<p>ASB 51.1 METALLURGICAL LITERATURE CLASSIFICATION</p>																																																			

Investigation of reactions in binary liquid systems. The tin bromide-ester murexide method of physical-chemical analysis. N. S. Kurnakov and K. B. Shternin. *Chemical analysis*. N. S. Kurnakov, N. S. Chernov, *ibid.*, No. 11, 1936, 467-92 (in German 492-3). Values of  $\alpha$  and  $\beta$  systems of SnBr<sub>4</sub> with Et formate, EtOAc, AcOMe, Me malate, Et malonate and Et ester of ethylmalonic acid, resp., were detd. The m. p., sp. heat and cond. of the SnBr<sub>4</sub> for AcEtBr system were detd. under various conditions. For binary liquid systems several compds. of varying degrees of dissociation are formed, the majority of these falling along the liquidus curve of the phase diagram. The max. of the viscosity isotherm occurs with the compd. of least decomposition. The min. in the cond. isotherm corresponds to the formation of a chem. compd., and a max. occurs only if there is appreciable dissociation. The max. of the sp. heat bears no relation to compd. formation. The maxima in the curve for  $\eta$  vs. the cond. correspond to the formation of compds. R. F. DeRight

ASB 31A METALLURGICAL LITERATURE CLASSIFICATION

1ST AND 2ND ORDER

PROCESS AND PROPERTIES INDEX

2

Ternary system *m*-toluenediamine-benzoic acid-salicylic acid. N. S. Kurnakov and E. B. Shaternina. *Ann. sector anal. phys. chim., Inst. chim. geo.* (U. S. S. R.) 13, 135-63(1940).—The compn. cond. and -viscosity curves of the systems *m*-C<sub>6</sub>H<sub>4</sub>(NH<sub>2</sub>)<sub>2</sub>-NaOH and -salicylic acid (I), and NaOH-I, at 90°, 100° and 125° confirm fu-sion-curve evidence of formation of 1:1 compds. in all the systems. Compa.-property diagrams based on these curves are presented for the ternary system, at 90°, 100° and 125°.

B. C. P. A.

ASA-SLA METALLURGICAL LITERATURE CLASSIFICATION

FROM DIVISION

FROM BOMILU

1ST AND 2ND ORDER

COMMON ELEMENTS

COMMON VARIABLES INDEX

*ca*

Preliminary report on experimental gasoline-storage tanks constructed according to the method of the Institute of General and Inorganic Chemistry of the Academy of Science, U.S.S.R. A. V. Nikolov and R. B. Shternina. *Zashchita Plomki na Solnko. Abad. Nauk S.S.S.R.* 1968, 75-80; cf. preceding abstracts.—Successful field trials were made on the use of lime-FeSO<sub>4</sub>-sand method for liquidproofing storage reservoirs for petroleum products.  
G. M. Kosolapoff

*20*

A.S.A. METALLURGICAL LITERATURE CLASSIFICATION

RESEARCH REPORTS

RESEARCH REPORTS

*C.A.*

*2*

Relubility in the system  $\text{CaCO}_3\text{-NaCl-CO}_2\text{-H}_2\text{O}$  at  
25°. H. B. Shtrikman and R. V. Prokaya. *Akad. Nauk  
S.S.S.R., Referaty, Otdel. Khim. Nauk* 1965, 32; cf.  
*C.A.* 40, 3871<sup>9</sup>.—Up to NaCl 2 moles/l. and  $p_{\text{CO}_2}$  1 atm.,  
the soly. of  $\text{CaCO}_3$  is expressed empirically by  $-\log a =$   
 $3.9081 - 0.3310 \log p_{\text{CO}_2}$  where  $a$  = activity. N. Thon

ASS-SLA METALLURGICAL LITERATURE CLASSIFICATION

FROM SYNOBOLM

SERIAL ONE ONE ONE

BULLET ONE

SEARCHED INDEXED

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1ST AND 2ND ORDERS													3RD AND 4TH ORDERS												
1ST AND 2ND ORDERS													3RD AND 4TH ORDERS												
PROCEDURES AND PROPERTIES INDEX																									
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<p>Solubility in the system <math>\text{CaCO}_3\text{-CaSO}_4\text{-NaCl-CO}_2\text{-H}_2\text{O}</math> at 25°. E. B. Shcherbina and R. V. Prokova (Institute of General and Inorganic Chemistry, Academy of Sciences of the U.S.S.R.). <i>Compt. rend. acad. sci. U.S.S.R.</i> 47, 33-5 (1945); <i>Doklady Akad. Nauk S.S.S.R.</i> 49, 34-6 (1945). — The solubilities of <math>\text{CaCO}_3</math> and <math>\text{CaSO}_4</math> in water contg. 0-2% NaCl at <math>\text{CO}_2</math> pressures of 0.0013-1.0 atm. are given. The data are expressed graphically in terms of the <math>\text{Ca}(\text{HCO}_3)_2</math> and <math>\text{CaSO}_4</math> contents of the solns. at equil. Use of the results to predict soln. of calcite and gypsum in river waters is indicated.</p> <p>J. H. Raley</p>																									
A.I.M.-S.L.A. METALLURGICAL LITERATURE CLASSIFICATION																									
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157 AND 158 CODES		PROCESS AND PROPERTIES INDEX	159 AND 160 CODES
<p>157 AND 158 CODES</p>		<p>PROCESS AND PROPERTIES INDEX</p> <p>2</p> <p>Calculation of the solubility of calcite in aqueous solutions. E. B. Shwarts. <i>Doklady Akad. Nauk S.S.S.R.</i> 30, 261-4 (1969); <i>cf. C.A.B.</i> 40, 2671<sup>1</sup>.—The equil. const. of <math>\text{CaCO}_3 + \text{CO}_2 + \text{H}_2\text{O} \rightleftharpoons \text{Ca}^{++} + 2\text{HCO}_3^-</math>, <math>K = [\text{Ca}(\text{HCO}_3)_2]/(0.0338 \rho)^{1/2}</math>, is ded. to <math>K = 0.000991</math> at 25°, <math>\rho</math> = partial pressure of <math>\text{CO}_2</math>, in atm., and the activity coeff. <math>f</math> is calcd. by <math>-\log f = 0.5001\sqrt{2}/\sqrt{\epsilon m}</math>, <math>= 0.5000\sqrt{2}/\sqrt{\epsilon m}</math> from data of 25°, and those of Frear and Johnston (<i>C.A.B.</i> 23, 636). Solub. of <math>\text{CaCO}_3</math> in <math>\text{NaCl}</math> aqns. from 0 to 2 <math>M</math>, <math>\text{CO}_2</math> from 0.00128 to 1 atm., at 25°, was calcd. from the activity <math>a</math> of <math>\text{Ca}(\text{HCO}_3)_2</math>, given, with the above value of <math>K</math>, by <math>-\log a = 3.8973 - 0.333 \log \rho</math>, and the activity coeff. <math>f</math> calcd. from the Debye-Hückel equation with the sum of the crystallographic radii of <math>\text{Na}^+</math> and <math>\text{Cl}^-</math>, 2.76 Å., taken for the mean effective diam. of the ions, i.e., <math>-\log f = 0.5002\sqrt{2}/\sqrt{\epsilon m}/(1 + 0.642\sqrt{2}/\sqrt{\epsilon m})</math>. The calcd. soly. curves, in mole fractions (<math>m</math>) <math>\text{Ca}(\text{HCO}_3)_2</math>, against <math>\text{NaCl}</math>, at <math>\rho = 0.00128</math>, 0.0002, 0.0406, and 0.0665 atm., fit very accurately the exptl. data. Examples of exptl. data: <math>\rho = 0.0002</math> atm., <math>\text{NaCl}</math> 0, 0.2007, 0.4005, 1.0223 <math>M</math>, <math>10^3 m = 5.91</math>, 8.45, 10.14, 13.80; <math>\rho = 0.0406</math> atm., <math>\text{NaCl}</math> 0, 0.1732, 0.8625, 1.8306, <math>10^3 m = 16.43</math>, 21.19, 28.71, 34.04.</p> <p>N. Thon</p>	<p>159 AND 160 CODES</p>
<p>ASH-SLA METALLURGICAL LITERATURE CLASSIFICATION</p>			
157 AND 158 CODES		159 AND 160 CODES	157 AND 158 CODES
157 AND 158 CODES		159 AND 160 CODES	157 AND 158 CODES

SHTERNINA, Ye. S.

— External components of the fusion diagram of the quaternary reciprocal system  $\text{NH}_4\text{Cl}$ ,  $\text{KCl}$ ,  $\text{NO}_3$ ,  $\text{H}_2\text{PO}_4$ . A. G. Bergman, V. P. Radtsimney, T. N. Mirosova, V. N. Evchenikova, E. S. Shternina, and M. A. Yulgar. *Izv. Akad. Nauk S.S.S.R. 15*, 157-69 (1947). — Studied were the quaternary systems of the 6 salts. In the system  $\text{KNO}_3$ - $\text{KCl}$  was observed the compd.  $\text{KNO}_3\text{KCl}$ , m.  $300^\circ$  (decomp.). This compd. had a polymorphic transformation at  $255^\circ$ . At the eutectic point  $320^\circ$  it contd. 8 mol. % of  $\text{KCl}$  and at the transition point  $17.4$  mol. %  $\text{KCl}$ . In the system  $\text{NH}_4\text{NO}_3$ - $\text{KNO}_3$  was observed the compd.  $2\text{NH}_4\text{NO}_3\text{KNO}_3$ , m.  $170^\circ$ . There were indications of the existence of a compd. richer in  $\text{KNO}_3$ . The eutectic of the double salt was at  $156.7^\circ$ , at which point it contd. 11.3 mol. % of  $\text{KNO}_3$ . At the transition point  $170.5^\circ$  it contd. 10.5 mol. % of  $\text{KNO}_3$ . In the eutectic,  $147^\circ$ , system  $\text{NH}_4\text{NO}_3$ - $\text{NH}_4\text{H}_2\text{PO}_4$  was observed a wide range of  $\text{NH}_4\text{H}_2\text{PO}_4$  solid solns. At the eutectic point this system contd. 12.5 mol. % of  $\text{NH}_4\text{H}_2\text{PO}_4$ . The eutectic of the system  $\text{KNO}_3$ - $\text{KH}_2\text{PO}_4$  was at  $244.5^\circ$  and contd. 29.2 mol. % of  $\text{KH}_2\text{PO}_4$ . In the system  $\text{KCl}$ - $\text{KH}_2\text{PO}_4$ , the eutectic point was at  $280^\circ$  when it contd. 5.5 mol. % of  $\text{KCl}$ . The system  $\text{NH}_4\text{Cl}$ - $\text{NH}_4\text{H}_2\text{PO}_4$  had a eutectic point at  $184^\circ$  when it contd. 12.2 mol. % of  $\text{NH}_4\text{Cl}$ .  $\text{NH}_4\text{H}_2\text{PO}_4$  and  $\text{KH}_2\text{PO}_4$  formed a continuous series of solid solns. The ternary system  $\text{NH}_4\text{NO}_3$ - $\text{NH}_4\text{H}_2\text{PO}_4$ - $\text{NH}_4\text{Cl}$  had a eutectic point at  $131.5^\circ$  when it contd.  $\text{NH}_4\text{NO}_3$  80.25,  $\text{NH}_4\text{H}_2\text{PO}_4$  5.75, and  $\text{NH}_4\text{Cl}$  14 mol. %. The eutectic point of the system  $\text{KNO}_3$ - $\text{KH}_2\text{PO}_4$ - $\text{KCl}$  was at  $235^\circ$  at which point it contd.  $\text{KNO}_3$  33.75,  $\text{KH}_2\text{PO}_4$  54.75, and  $\text{KCl}$  7.5 mol. %. Since within this system



the quaternary reciprocal system  $\text{NH}_4, \text{K} \parallel \text{Cl}, \text{NO}_3, \text{H}_2\text{PO}_4$ , *Ibid.* 200-33.—The fusion diagram of the system was studied up to  $180^\circ$  in the region adjacent to  $\text{NH}_4\text{NO}_3$ . Within this region were crystd.  $\text{NH}_4\text{NO}_3$ ,  $2\text{NH}_4\text{NO}_3 \cdot \text{KNO}_3$ ,  $\text{KNO}_3$ ,  $\text{NH}_4(\text{K})\text{Cl}$ , and  $\text{NH}_4(\text{K})\text{H}_2\text{PO}_4$ . These spaces made contact at the eutectic point approx.  $123^\circ$  and at the transition point approx.  $134^\circ$ . At the eutectic point the approx. compn. was  $\text{NH}_4$  90.5,  $\text{K}$  9.5,  $\text{Cl}$  13.75,  $\text{H}_2\text{PO}_4$  3.75,  $\text{NO}_3$  82.50 ion %, and the solid phases  $\text{NH}_4(\text{K})\text{Cl} + \text{NH}_4(\text{K})\text{H}_2\text{PO}_4 + \text{NH}_4\text{NO}_3 + 2\text{NH}_4\text{NO}_3 \cdot \text{KNO}_3$ . At the transition point the approx. compn. was  $\text{NH}_4$  86.0,  $\text{K}$  14.0,  $\text{Cl}$  13.25,  $\text{H}_2\text{PO}_4$  4.80,  $\text{NO}_3$  81.95 ion %, and the solid phases  $2\text{NH}_4\text{NO}_3 \cdot \text{KNO}_3 + \text{NH}_4(\text{K})\text{Cl} + \text{NH}_4(\text{K})\text{H}_2\text{PO}_4 + \text{KNO}_3$ . Also studied was the effect of addn. of  $\text{NaCl}$  and  $\text{KH}_2\text{PO}_4$  on the m.p. Addn. of  $\text{NaCl}$  lowered the quaternary eutectic point,  $123^\circ$ ,

by  $9^\circ$ . Addn. of  $\text{KH}_2\text{PO}_4$  lowered the eutectic point,  $112^\circ$ , of the ternary system  $\text{NH}_4, \text{Na} \parallel \text{Cl}, \text{NO}_3$  by  $9^\circ$ .

M. Hoesch

SHTERNINA, Ye. B.

62T3

USSR/Chemistry - Gypsum - Solubility  
Chemistry - Potassium Sulfate

Apr 1948

"Trial Computation of the Solubility of Potassium Sulfate in a Water Solution of Sodium Chloride," E. B. Shternina, 4 pp

"Dok Akad Nauk SSSR, Nova Ser" Vol LX, No 2

Data on the solubility of gypsum in water solutions of salt, and carbon dioxide at 25° temperature, from which it was simple to determine the coefficient of activity of potassium sulfate. Debye's and Haackel's formulas were used to determine this activity. Submitted by Academician G. G. Urazov, 13 Feb 1948

62T3

CA

2

Solubility of gypsum in salt solutions. R. B. Shernina. *Izvest. Sektora Fiz.-Khim. Anal., Inst. Obshch. i Neorg. Khim., Akad. Nauk S.S.S.R.* 17, 351-59 (1940); cf. *C.A.* 40, 3971<sup>+</sup>; 43, 6000<sup>+</sup>.—The system investigated was  $\text{CaSO}_4$ - $\text{NaCl}$ - $\text{CO}_2$ - $\text{H}_2\text{O}$  at 25°. The concn. of  $\text{NaCl}$  varied from 0 to satn. The partial pressure of  $\text{CO}_2$  did not exceed 1 atm. Every 2-3 days the liquid phase was analyzed for  $\text{Ca}^{++}$ ,  $\text{SO}_4^{--}$ , and  $\text{Cl}^-$ . This was continued until 3 consecutive analyses differed by no more than 0.01 g. of  $\text{CaSO}_4$  per 1 l. of soln.  $\text{CO}_2$  had no effect on the soly. of  $\text{CaSO}_4$ . The soly. of  $\text{CaSO}_4$  was calcd. by means of the Debye-Hückel equation (*C.A.* 17, 2065; 18, 3514) in which the values of consts. were taken from Birge (*C.A.* 36, 1221<sup>+</sup>) and Wensel (*C.A.* 33, 4635<sup>+</sup>). As the concn. of  $\text{NaCl}$  increased from 0 to 317.226 g./l. (0-5.427 mole./l.), the soly. of  $\text{CaSO}_4$  rose from 2.063 (0.01530 mol./l.) to a max. of 7.236 g./l. (0.03315 mol./l.) and then dropped to 5.816 g./l. (0.04272 mol./l.). The max. occurred at a  $\text{NaCl}$  concn. of 138.756 g./l. The exptl. curves giving the relation between the mean effective diam. of ions ( $b$ ) and concn. can generally be expressed by

$b = b_0 + (b_{\text{sat}} / \sum c_{\text{sat}}^{\text{sat}}) \sum c_{\text{sat}}^{\text{sat}}$ , where  $b_0$  is the av. crystallographic diam. of  $\text{Ca}^{++}$  and  $\text{SO}_4^{--}$  (= 3.11 Å.),  $b_{\text{sat}}$  is the av. crystallographic diam. of the ions of the added salt,  $\sum c_{\text{sat}}^{\text{sat}}$  is the ionic strength of a satd. soln. of the added salt, and  $\sum c_{\text{sat}}^{\text{sat}}$  is the ionic strength of the given soln. of the added salt. M. Hoesch

~~SHTERNINA, M.B.~~

D.I. Mendeleev theory on solutions. Izv. Sekt. fiz. khim. anal.  
18:231-246 '49. (MIRA 11:4)

1. Institut obshchey i neorganicheskoy khimii im. N.S. Kurnakova  
AN SSSR.

(Mendeleev Dmitrii Ivanovich, 1834-1907)  
(Solution (Chemistry))

SHTERNINA, E. B.

Solubility of calcite in the presence of carbon dioxide and sodium chloride. E. B. Shternina and E. V. Prolova (N. S. Kurnakov Inst. Gen. Inorg. Chem., Acad. Sci., U.S.S.R., Moscow). *Izvest. Sektsiya Fiz.-Khim. Anal. Inst. Obshchey Neorg. Khim., Akad. Nauk S.S.S.R.* 21, 271-87(1952).—Ice land spar crushed to  $+0.5$  mm. was used, and the expts. were made at  $25 \pm 0.05^\circ$ . In the system  $\text{CaCO}_3\text{-NaCl-CO}_2\text{-H}_2\text{O}$  equil. was attained only after 1-2 months. To get the soly. of a sparingly sol. salt are required the activity ( $a$ ) and the coeff. of activity ( $\gamma$ ) of this salt.  $a$  was detd. in dil. solns. and was calcd. from the Debye-Hückel equation assuming that the size of ions increases directly with the concn. of the sol. salt. This relation can be expressed by  $b = b_s + (b_w / \sum m_i Z_i^2) \sum m_i s_i$ ,  
sat.

where  $b$  is the av. size of ions in the system,  $b_s$  and  $b_w$  are the av. crystallographic sizes of ions of the sparsely ( $s$ ) and well ( $w$ ) sol. salts,  $\sum m_i Z_i^2$  is the ionic strength of the added well-sol. salt in satd. soln., and  $\sum m_i s_i$  is the ionic strength of the given soln. of the added well-sol. salt. The soly. of calcite and gypsum in the presence of NaCl had a max. but the corresponding  $\gamma$  had a min. If it were assumed that the av. distance between ionic centers is independent of concn., then the soly. of gypsum would be constantly increasing and  $\gamma$  constantly decreasing with increasing concn. of NaCl. In the case of gypsum the max. on the soly. curve corresponds to the min. on the  $\gamma$  curve which in turn is detd. by the increase in the av. distance between ions with increasing concn. of NaCl. In the case of calcite, the situation is more complicated, since the  $a$  of  $\text{Ca}(\text{HCO}_3)_2$  does not remain const. but changes with  $\text{CO}_2$  content according to  $a = 0.0213 [\text{CO}_2]^{1/2}$ . Also NaCl has a salting-out effect on  $\text{CO}_2$ . Thus, NaCl on the one hand increases the soly. of  $\text{CaCO}_3$ , even as of  $\text{CaSO}_4$ , and on the other hand it lowers the soly. of  $\text{CaCO}_3$  by decreasing the soly. of  $\text{CO}_2$ . M. Hosh

SHTERNINA, E. B.

①

E. T. R.  
June 1954  
Chemistry-Physical

7715 Calculation of Solubility of Calcite in Aqueous Solutions. E. B. Shternina. Henry Brucher, Altadena, Calif., *Translation no. 3182, 8 p.* (From *Doklady Akademii Nauk SSSR*, v. 50, 1954, p. 261-264.)

Application of quantitative thermodynamics and modern solution theory to construction of solubility diagrams. Tables, graph. 9 ref.

SHTERNA. E.B.

✓ Maximums on the solubility isotherms for calcium sulfate and calcium carbonate. B. B. Shternina. *Zhur. Neorg. Khim.* 2, 933-7(1957). — The max. observed on the soly. isotherms for the system  $\text{CaCO}_3\text{--CaSO}_4\text{--NaCl--CO}_2\text{--H}_2\text{O}$  (C.A. 45, 4126g; 48, 8224a) was considered theoretically; it was assumed that the activity of the dissolved salt is const., and the Debye-Hückel equations were applied. A comparison of exptl. and theoretical data verifies the assumptions which were made. J. Roytar-Leach

3

1-4E4J

1-4E3d

NS

SHTERNINA, E.B.

Solubility of poorly soluble salts in aqueous solutions of  
electrolytes. Itogi nauki: Khim.nauki 4:55-60 '59.  
(MIRA 13:4)

(Electrolytes) (Solubility) (Salts)



SHTERNINA, E.B.

Solubility in complex salt systems. Biul. MOIP. Otd. geol. 34  
no.5:158-159 S-0 '59. (MIRA 14:6)  
(Solubility)

5.4120

78055  
SOV/62-60-1-1/37

AUTHOR: Shternina, E. B.

TITLE: Concerning Solubility of Salts in Aqueous Solutions of Electrolytes

PERIODICAL: Izvestiya Akademii nauk. Otdeleniye khimicheskikh nauk, 1960, Nr 1, pp 3-8 (USSR)

ABSTRACT: Determination of solubility of salts in complex systems, like concentrated aqueous solutions containing ions of the same salt or ions of other salts, was studied. Different methods and formulae for calculating solubilities are reviewed. The author states that the various known formulae are not applicable to the concentrated and complex systems. A new method of determination of solubility is proposed. Solubility ( $x$ ) is considered as a product of solubility coefficient ( $\beta$ ) and activity ( $a$ )

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$$x = \beta a \quad (5)$$

Concerning Solubility of Salts in Aqueous  
Solutions of Electrolytes

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$\alpha$  is a constant. It was assumed that the solubility coefficient rises with electrostatic interaction of ions and falls with decreasing volume concentration of water

$$\beta = \frac{z_1 z_2 \dots \sqrt{\sum m_i z_i^2}}{\frac{D_1 - D}{C}} \quad (6)$$

where  $z_1 z_2$  are the valences of the solid ions;

$\sqrt{\sum m_i z_i^2}$  is the ionic strength of solution;

$D_1$  is the number of moles of water in 1 liter of pure water,  $D_1 = 55.3434$  at  $25^\circ$ ;  $D$  is moles of water in 1 liter of solution.

$$D = \frac{55.506 d}{1000 + g}, \quad (7)$$

where 55.506 is the number of moles of water in 1,000 g of water;  $d$  is density of the solution;  $g$  is total

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Concerning Solubility of Salts in Aqueous  
Solutions of Electrolytes

78055  
SOV/62-60-1-1/37

weight (grams) of all dissolved compounds per 1,000 g of water;  $C$  is the sum of moles of all dissolved compounds in 1 liter of solution,  $C = \sum c_i$ , where  $c_i$  is moles of compound  $i$  per liter of solution.

$$c_i = \frac{1000 m_i d}{1000 + g} \quad (8)$$

where  $m_i$  is moles of compound  $i$  per 1,000 g of water. Solubilities calculated according to the above method are confirmed by experimental data. There are 20 references, 2 U.S., 1 U.K., 2 Indian, 4 German, and 11 Soviet. The U.S. and U.K. references are: R. A. Robinson, R. H. Stokes, Electrolyte Solutions, London, 1955; T. H. Gronwall, V. K. La Mer, L. J. Greiff, J. Phys. Chem 35, 2245 (1931); E. Wicke, M. Eigen, J. Phys. Chem., 58, 702 (1954).

ASSOCIATION:

N. S. Kurnakov Institute of General and Inorganic Chemistry, Academy of Sciences USSR (Institut obshchey i neorganicheskoy khimii imeni N. S. Kurnakova Akademii nauk SSSR)

SUBMITTED:

May 28, 1959

Card 3/3

SHTERNINA, E.B.; FROLOVA, Ye.V.

Removal of ballast carbonates from Kara-Tau phosphorite ores.  
Zhur.prikl.khim. 35 no.4:751-756 Ap '62. (MIRA 15:4)

1. Institut obshchey i neorganicheskoy khimii imeni N.S.Kurnakova  
AN SSSR.  
(Kara-Tau--Phosphorites) (Carbonates)

SHTERNINA, E.B.; RYKOVA, G.A.

Minimum on solubility isotherms. Zhur. neorg. khim. 10 no.9:2152-  
2155 S '65. (MIRA 18:10)

SALGANIK, R.I., kand.biolog.nauk; SHTERNISHIS, Yu.S. (Novosibirsk)

Use of deoxyribonuclease in suppurative processes in the lungs.  
Klin.med. 40 no.6:95-100 Je '62. (MIRA 15:9)

1. Iz laboratorii nukleinovykh kislot i nukleoproteidov (zav. -  
kand.biol.nauk R.I. Salganik) Instituta tsitologii i genetiki  
Sibirskogo otdeleniya AN SSSR.  
(LUNGS---DISEASES) (DEOXYRIBONUCLEASE)

L 62811-65

ACCESSION NR: AP5016387

UR/0120/65/000/003/0129/0131  
621.375.2.4

AUTHOR: Shternov, A. A.

TITLE: Tunnel diode pulse amplifier

SOURCE: Pribery i tekhnika eksperimenta, no. 3, 1965, 129-131

TOPIC TAGS: tunnel diode amplifier, semiconductor pulse amplifier, pulse amplifier

ABSTRACT: A TD pulse amplifier with a 340-Mc amplification area has been studied theoretically and experimentally. Maximum amplification was obtained by connecting the diode in parallel to the load as shown in Fig. 1 of the Enclosure. The gain is 1.35 and the rise time depends essentially on gain. For practical applications the amplified signal should be blocked from the d-c bias source. Orig. art. has: 9 formulas and 2 figures. [08]

ASSOCIATION: Nauchno-issledovatel'skiy radiofizicheskiy institut pri GGU (Scientific Research Institute of Radio Physics at GGU)

SUBMITTED: 17May64

ENCL: 01

SUB CODE: EC

NC REF SOV: 001

OTHER: 001

ATD PRESS: 4056

Cord 1/2



L 62811-65

ACCESSION NR: AP5016387

ENCLOSURE: 01

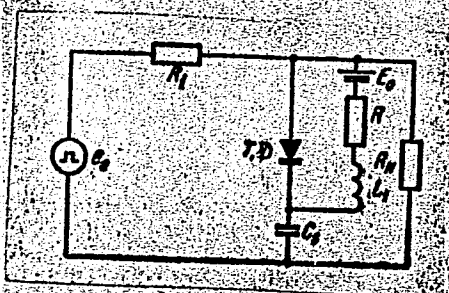


Fig. 1. Tunnel diode pulse amplifier

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ACCESSION NR: AP4029460

S/0108/64/019/004/0046/0051

AUTHOR: Lesin, Yu. S. (Active member); Shternov, A. A. (Active member)

TITLE: Noncoherent exponential-weight storage of packets of pulse signals with nonsquare envelopes

SOURCE: Radiotekhnika, v. 19, no. 4, 1964, 46-51

TOPIC TAGS: pulse signal, pulse packet, nonsquare envelope, pulse packet storage, radar detection

ABSTRACT: An approximate calculation is made of the threshold signal-to-noise ratio for two nonsquare cases of pulse-packet envelopes. For triangular and sinusoidal envelopes, formulas are developed for the probability of correct radar detection; also, the threshold signal-to-noise ratio vs. pulse number per packet curves are presented; a square-law exponential-weight packet storage is assumed. It is proven that, with a constant maximum signal amplitude and pulse

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ACCESSION NR: AP4029460

number per packet, a change in the envelope shape -- from triangular to sinusoidal or from sinusoidal to square -- results in a reduction of the threshold signals. This may be explained by an increase in the energy of the packet. Orig. art. has: 3 figures and 16 formulas.

ASSOCIATION: Nauchno-tekhnicheskoye obshchestvo radiotekhniki i elektrosvyazi  
(Scientific and Technical Society of Radio Engineering and Electrocommunication)

SUBMITTED: 12Feb63

DATE ACQ: 30Apr64

ENCL: 00

SUB CODE: EC, DC

NO REF SOV: 005

OTHER: 000

Card 2/2

AYZEN/ERG, D.Ye.; BELEVTSSEV, Ya.M.; BORDUNOV, I.N.; BORISENKO, S.T.;  
BULKIN, G.A.; GORLITSKIY, B.A.; DOVGAN', M.N.; ZAGORUYKO,  
L.G.; KAZAKOV, L.R.; KALYAYEV, G.I.; KARASIK, M.A.; KACHAN,  
V.G.; KISELEV, A.S.; LAGUTIN, P.K.; LAZARENKO, Ye.K.;  
LAZARENKO, E.A.; LAPITSKIY, E.M.; LAPCHIK, F.Ye.; LAS'KOV,  
V.A.; LEVENSHTeyN, M.L.; MALAKHOVSKIY, V.F.; MITKEYEV, M.V.;  
PRUSS, A.K.; SKARZHINSKIY, V.I.; SKURIDIN, S.A.; SOLOV'YEV,  
F.I.; STRYGIN, A.I.; SUSHCHUK, Ye.G.; TEPLITSKAYA, N.V.;  
FEDYUSHIN, S.Ye.; FOMENKO, V.Yu.; SHKOLA, T.N.; SHTERNOV,  
A.G.; YAROSHCHUK, M.A.; ZAVIRYUKHINA, V.N., red.

[Problems of metallogeny in the Ukraine] Problemy metallo-  
genii Ukrainy. Kiev, Naukova dumka, 1964. 254 p.  
(MIRA 18:1)

1. Akademiya nauk URSR, Kiev. Instytut geologichnykh nauk.

SHTERNOV, M. M.

SHTERNOV, M. M. -- "Systems of Metal-Drawing Grooves." Min Higher Education USSR. Ural Polytechnic Inst imeni S. M. Kirov. Sverdlovsk, 1955.  
(Dissertation for the Degree of Candidate of Technical Sciences.)

SO: Knizhnaya Letopis', No 5, Moscow, Feb 1956

KOZHEVNIKOV, V.P., inzhener; BAKHTINOV, B.P., inzhener; MEREKIN, S.V.,  
inzhener; ~~SHTERNOV~~, M.M., inzhener; GRITSUK, N.F., inzhener.

Turn-over rollers for continuous billet mills. Stal' 15 no.1:54-58  
Ja '55. (MIRA 8:5)

1. Magnitogorskiy metallurgicheskiy kombinat.  
(Rolling-mill machinery)

SHTERNOV, M.M., kandidat tekhnicheskikh nauk.

Rolling special channel iron with wide flanges. Stal' 17 no.1:90-  
91 Ja '57. (MIRA 10:3)

1. Magnitogorskiy metallurgicheskiy kombinat.  
(Rolling (Metalwork))

BAKHTINOV, B.P.; FURMAN, Ya.B; SHTERNOV, M.M.

"Wear of iron mill rolls" by A.P. Chekmarev, R.A. Moshkovtsev.  
Reviewed by B.P. Bakhtinov, Ia.B.Furman, M.M. Shternov. Stal'  
17 no.1:95-96 Ja '57. (MLRA 10:3)

1. Magnitogorskiy metallurgicheskiy kombinat.  
(Rolls (Iron mills))

(Chekmarev, A.P.) (Moshkovtsev, R.A.)



130-1-12/17

AUTHORS: Shternov, M.M., Candidate of Technical Sciences, and  
Gritsuk, N.F., Engineer.

TITLE: New Designs of Roller Rolling Equipment (~~Novyye~~ konstruktsii  
rolikovoy valkovoy armatury)

PERIODICAL: Metallurg, 1958, No.1, pp. 23 - 28 (USSR)

ABSTRACT: Some designs of guide blocks developed in connection with the increasing speed of rolling are described. The equipment described was designed by the roll-pass department of the Magnitogorsk Metallurgical Combine in collaboration with mill operators for the automatic 300 mill. The mill is intended to roll small channels, angles, rounds, squares and strip. It is of the cross-country type with ten working stands, the rolling speed in the finishing stands being 9-11 m/second with a billet weight of 200-500 kg. The article gives details of the front and back guide blocks for some of the products and discusses the advantages of these roller types over the friction types. The dimension of the guide blocks in relation to each other and the stand dimensions have been selected to minimise the amount of equipment required for the range of products rolled. The designs provide for the rapid changing of blocks and for mechanized movement along the barrel of the roll, and are based on the principle of the separation of parts needing adjustment during work from the fixing elements. It is

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130-1-12/17

New Designs of Roller Rolling Equipment

suggested that the rollers should be made of ordinary engineering steels and hard-faced with wear-resistant alloys and that high-grade textolite is a suitable bearing material for the finishing and pre-finishing stands with rolling speeds over 6-6.5 m/sec. The possibility of using high-chromium iron inserts instead of the rollers at the inlet to the roughing stands is briefly mentioned. In the design of the equipment, Soviet and foreign practical experience is said to have been utilized. There are 4 figures.

ASSOCIATION: Magnitogorsk Metallurgical Combine (Magnitogorskiy metallurgicheskiy kombinat)

AVAILABLE: Library of Congress

Card 2/2

SOV/135-59-5-14/31

AUTHOR: Shternov, M.M., Candidate of Technical Sciences  
TITLE: A Modern Continuous Billet Mill (Sovremennyy nepreryvno-  
zagotovochnyy stan)  
PERIODICAL: Stal', 1959, Nr 5, pp 429 - 432 (USSR)

ABSTRACT: The relative merits of continuous billet mills with only horizontal stands and with alternating horizontal and vertical stands are discussed. The author is in favour of mills of the first type, stressing their lower capital and maintenance costs, the possibility of using longer rolls with many rolling lines, ease of transfer from one profile to another, the possibility of transferring in one rolling line from rectangular-square passes to rhombic-square passes, etc. It is pointed out in the editorial notes that some of the author's concepts are based on experience of operation of mills with horizontal stands only and can be contested. As the problem is of great importance,

Card1/2

A. Modern Continuous Billet Mill

SOV/153-59-5-14/31

discussion on the subject is invited.

There are 3 figures.

ASSOCIATION: Magnitogorskiy metallurgicheskiy kombinat  
(Magnitogorsk Metallurgical Combine)

Card 2/2